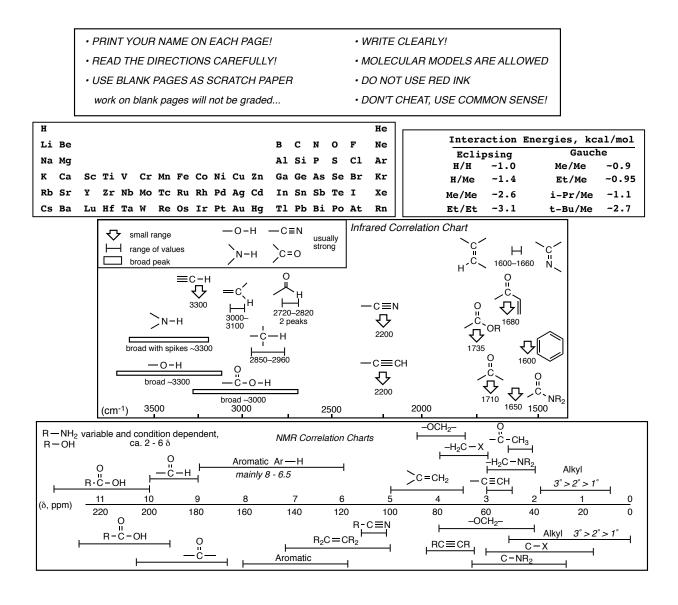
CHEM 234, Spring 2019	Midterm #3	lan R. Gould
COMPLETE THIS SECTION : Up to TWO POINTS	will be removed for incorrect/missing information	n!
PRINTED FIRST NAME	PRINTED LAST NAME	
Person on your LEFT (or Empty	or <b>Aisle</b> )	
Person on your <b>RIGHT</b> (or <b>Empty</b> or	Aisle)	
Class you are REGISTERED FOR (onground of	or hybrid)	
The room where most students will take the test class, i.e. LS A-191 for onground and PS H-152 fo		

## \*\*YOU ARE NOT ALLOWED TO TAKE SPARE COPIES OF THIS EXAM FROM THE TESTING ROOM\*\*



Midterm #3

## YOU MUST COMPLETE THIS PAGE WITH YOUR NAME (EVEN THOUGH YOU ALREADY DID THIS ON THE COVER PAGE) AND ALSO GIVE YOUR ASU OR POSTING ID NUMBER WE NEED THIS NUMBER BECAUSE YOU WOULDN'T BELIEVE THE NUMBER OF STUDENTS WHOSE NAMES WE CAN'T READ!

PRINTED FIRST NAME	PRINTED LAST NAME		ASU ID or Posting ID	
Points by question				
	1	_/13		
	2	_/22		
	3	_/42		
	4	_/40		
	5	_/36		
	6	_/22		

Points Removed for cover errors \_\_\_\_/2

Extra Credit\_\_\_\_/5

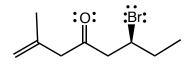
Total (incl Extra) /175+5

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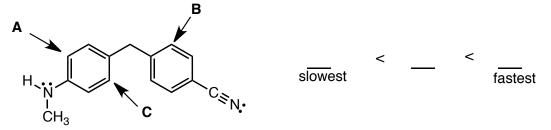
NAME

- 3 -

Question 1 (13 pts). Give the IUPAC name for the following compound. Be sure to use cis/trans, E/Z or R/S where appropriate.



Question 2 (22 pts). Rank in order of increasing rate of electrophilic aromatic substitution at the carbons indicated by the arrows. Give a BRIEF explanation.



Extra Credit (5 pts). Health problems associated with the use of Agent Orange defoliant in the Vietnam war are thoight to be associated with the presence of an impurity produced when Agent Orange was synthesized. This impurity had what kind of chemical structure?

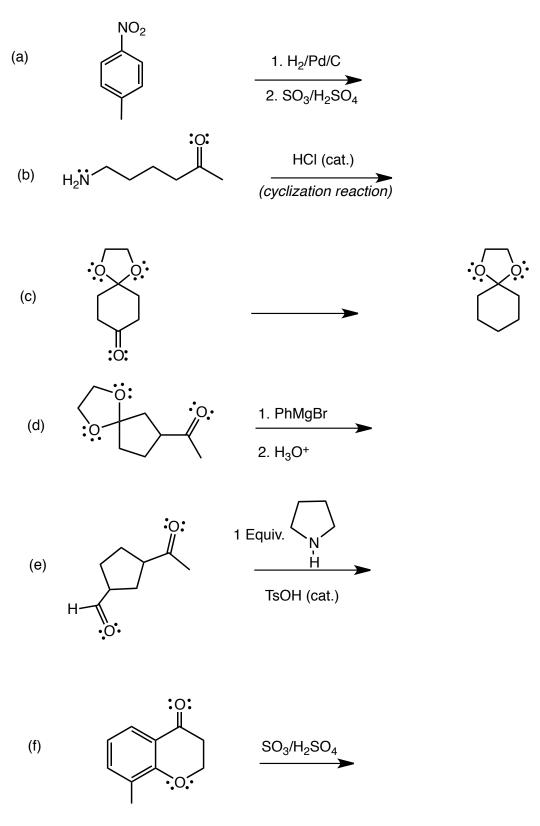
Chlorinated Aldehyde

Chlorinated Ketone

**Chlorinated Aromatic** 

- 4 - NAME

Question 3 (42 pts.) Provide the missing major organic products or reagents/conditions as appropriate, you can IGNORE stereochemistry in these problems



- 5 -

NAME

Question 4 (40 pts.) Synthesize the (target) molecules on the right from the starting molecules the left. this can not be done in one reaction. Give reagents and conditions and the intermediate molecules at each step. Do not show any mechanisms or transient intermediates. If other isomers are formed at any step then you need to indicate this but you do not need to draw their structures.

 $NO_2$ .CO<sub>2</sub>H a) SO<sub>3</sub>H



NAME

Question 5 (36 pts.) For the following TWO reactions a) and b):

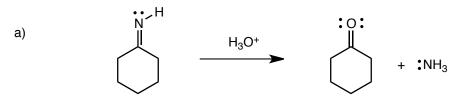
1) Give complete cuerved arrow-pushing mechanisms

2) Indicate the lewis acid/base for each INTERmolecular step (LB or LA) and whether they are also Brønsted bases/acids (LB/BB or LA/BA)

- 6 -

3) Show where every proton comes from and goes to (i.e., no +H<sup>+</sup> or -H<sup>+</sup>)

4) DRAW ALL RELEVANT RESONANCE CONTRIBUTORS FOR THE INTERMEDIATES



the ammonia will protonate under the reaction conditions, but we will ignore that here

Give the number of steps in your mechanism for reaction a) \_\_\_\_\_



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 $\cap$ 





Give the number of steps in your mechanism for reaction b) \_\_\_\_\_

NAME

Question 6 (22 pts.) For the following TWO reactions a) and b):

1) Give complete cuerved arrow-pushing mechanisms

2) Indicate the lewis acid/base for each INTERmolecular step (LB or LA) and whether they are also Brønsted bases/acids (LB/BB or LA/BA)

- 7 -

3) Show where every proton comes from and goes to (i.e., no +H<sup>+</sup> or -H<sup>+</sup>)

4) DRAW ALL RELEVANT RESONANCE CONTRIBUTORS FOR THE INTERMEDIATES

