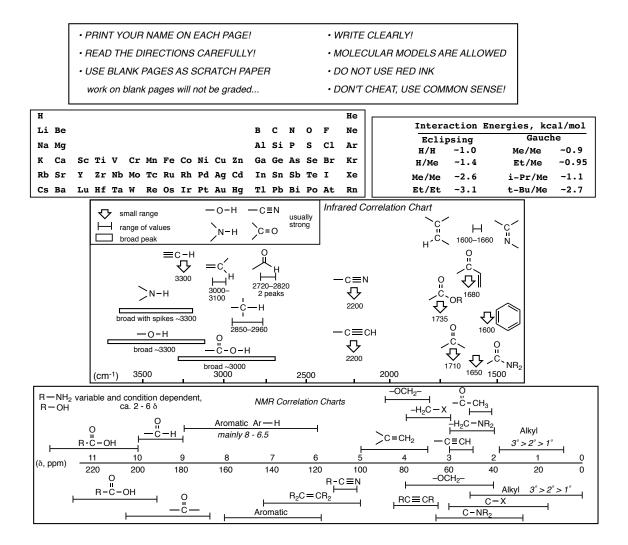
CHEM 234, Spring 2018 Midter	Midterm #3	
COMPLETE THIS SECTION : Up to TWO POINTS will be	removed for incorrect/missing information! —	
PRINTED FIRST NAME	PRINTED LAST NAME	
Person on your LEFT (or Empty or Aisl	e)	
Person on your RIGHT (or Empty or Aisle)	·	
Class you are REGISTERED FOR (onground or hybri	d)	
The room where most students will take the test for you class, i.e. LS A-191 for onground and PS H-152 for hybrid	ur ()	

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CHEM 234, Spring 2018

Midterm #3

YOU MUST COMPLETE THIS PAGE WITH YOUR NAME (EVEN THOUGH YOU ALREADY DID THIS ON THE COVER PAGE) AND ALSO GIVE YOUR ASU OR POSTING ID NUMBER WE NEED THIS NUMBER BECAUSE YOU WOULDN'T BELIEVE THE NUMBER OF STUDENTS WHOSE NAMES WE CAN'T READ!

PRINTED FIRST NAME	PRINTED LAST NAME		ASU ID or — Posting ID ———	
	Points by qu	estion		
	1	/13		
	2	/22		
	3	/40		
	4	/40		
	5	/40		
	5	/20		

Points Removed for cover errors ____/2

Extra Credit____/5

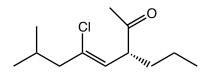
Total (incl Extra)____/175+5

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CHM 234, Spring 2018, Midterm #3

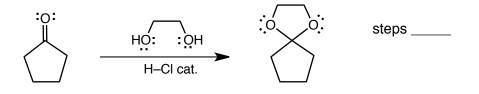
- 2 - NAME

Question 1 (13 pts). Give the IUPAC name for the following compound. Be sure to use cis/trans, E/Z or R/S where appropriate.



Question 2 (22 pts.) Give a arrow-pushing mechanism for the following reaction, indicate the Lewis and Bronsted acids/bases for each intermoelcular step (LB, LB, BB, BA)

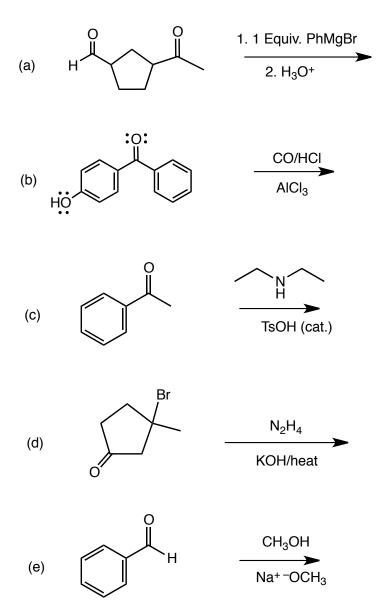
- $\boldsymbol{\cdot}$ Show all resonance contributors for the intermediates
- Show where all protons come from and go to (no +H⁺/-H⁺)
- $\boldsymbol{\cdot}$ Give the number of steps in your mechanism



CHM 234, Spring 2018, Midterm #3

- 3 - NAME

Question 3 (40 pts.) Provide the missing major organic products or reagents/conditions as appropriate, you can IGNORE stereochemistry in these problems



Extra credit question (5 pts).β-carotene is synthesized using which reaction?ClemmensonGrignardWittigAldol

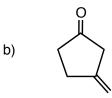
CHEM 234, Spring 2018 MIDTERM #3

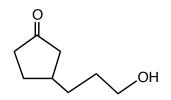
NAME

Question 4 (40 pts.) Synthesize the (target) molecules on the right from the starting molecules the left. this can not be done in one reaction. Give reagents and conditions and the intermediate molecules at each step. Do not show any mechanisms or transient intermediates. If other isomers are formed at any step then you need to indicate this but you do not need to draw their structures.

- 4 -

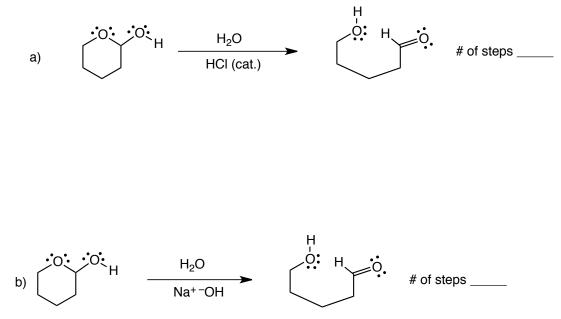
CO₂H a) O_2N Br





CHM 234, Spring 2018, Midterm #3 -5- NAME Question 5 (40 pts.) Give a complete arrow-pushing mechanism for the following reactions, indicate the Lewis and Bronsted acids/bases for each intermolecular step (LB, LB, BB, BA) • Show all resonance contributors for the intermediates

- · Show where all protons come from and go to (no +H+/-H+)
- · Give the number of steps in your mechanism



Question 6 (20 pts). Rank the following reactions in order of increasing rate. Give a BRIEF explanation that includes the terms Lewis acidity/basicity **OR** nucleophilicity/electrophilicity (almost all points for the explanation, not getting the order correct)

