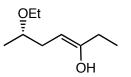
CHEM 234, Spring 2009	Se	econd Mi	idterm			Ian R. Gould
PRINTED PRINTEL					ASU ID c	or
FIRST NAME LAST	「NAME_				— Posting II	D
Person on your LEFT (or Aisle)			Person on your RIGHT (or Aisle)			
• PRINT YOUR NAME ON EACH PAGE!] 1				9	/8
• READ THE DIRECTIONS CAREFULLY!	2	/	10		10	/12
· USE BLANK PAGES AS SCRATCH PAPER	3	/	9		11	/12
• USE BLANK PAGES AS SCHATCH PAPEH work on blank pages will not be graded	4	/9	9		12	/25
•WRITE CLEARLY!	5	/	18			
• MOLECULAR MODELS ARE ALLOWED	6	/2	25			
• MOLECULAR MODELS ARE ALLOWED • DO NOT USE RED INK		/				
	8	/	1 2			
• DON'T CHEAT, USE COMMON SENSE!	Extra	a Credit_	/	5	Total (incl E	xtra)/175+5
Н				He	Interacti	on Energies, kcal/mol
Li Be	В	C N	O F	Ne	Eclipsir	ng Gauche
Na Mg	A	l Si P	s ci	l Ar	Н/Н ~1	.0 Me/Me ~0.9
K Ca Sc Ti V Cr Mn Fe Co Ni Cu	u Zn Ga	a Ge As	Se Bı	r Kr	H/Me ~1	.4 Et/Me ~0.95
Rb Sr Y Zr Nb Mo Tc Ru Rh Pd Ag	g Cđ Ir	n Sn Sb	Te I	Xe		.6 i-Pr/Me ~1.1
Cs Ba Lu Hf Ta W Re Os Ir Pt Au	u Hg T]	l Pb Bi	Po At	t Rn	Me/Et ~2	.9 t-Bu/Me ~2.7
		Infrared C		`c_′	H ∑c∕	Approximate Coupling Constants, J (Hz), for ¹ H NMR Spectra
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	2 — C 2 500	2200 2200 2200 2200 2200 2200	00	0 0 0 0 0 0 0 0 0 0 0 0 0 0	0 1600 0 1650 1500	$\begin{array}{c c} H & H \\ -C - C - & -7 \\ \hline \\ -C - C - & -7 \\ \hline \\ -C - C - & -7 \\ \hline \\ H \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$
amine $R - NH_2$ variable and condition alcohol $R - OH$ dependent, ca. 2 - 6 δ						
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	7 140 	$\frac{6}{120}$ R R ₂ C=CR ₂	⊢ 5 100 −C≡N ⊢ –) = c= (()	-OCH₂- RC≡CR	$\begin{array}{c c} \hline & & \\ \hline & & \\ \hline & & \\ \hline \\ \hline \\ H \\ \hline \\ & 2 \\ \hline \\ \hline \\ & 2 \\ \hline \\ & 0 \\ \hline \\ & 1 \\ \hline \\ & 0 \\ \hline \\ & 1 \\ \hline \\ & 0 \\ \hline \\ \hline \\ & 0 \\ \hline \\ \hline \\ \\ & 0 \\ \hline \\ \hline \\ \\ \hline \\ \hline \\ \\ \hline \\ \hline \\ \hline \\ \hline \\$
	Aromatic	<u> </u>			⊢	

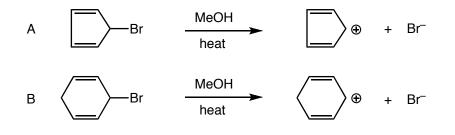
NAME

Question 1 (10 pts.) Give the IUPAC name for the following compound. Be sure to use cis/trans, E/Z or R/S where appropriate.

- 2 -



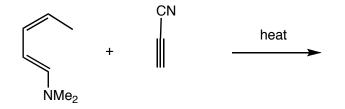
Question 2 (10 pts) Indicate which of the following two reactions, A or B would be faster. GIVE AN EXPLANATION THAT INCLUDES A DISCUSSION OF THE EXOTHERMICITY OR ENDOTHERMICITIES OF THE REACTIONS (only 1 point for the correct answer, 9 pts for the explanation)



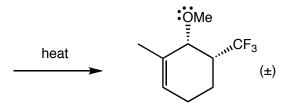
NAME

Question 3 (9 pts) Give the product of the following Diels-Alder reaction. Indicate both the relative and the absolute stereochemistry

- 3 -



Qestion 4 (9 pts) Give the diene and dienophile that react to give the provided structure in a Diels-Alder reaction



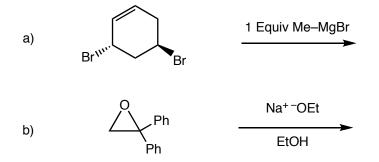
Extra Credit Question (5 pts.) In which kind of biomolecule can a photochemical 2 + 2 cycloaddition reaction take place

DNA phospholipid carbohydrate protein

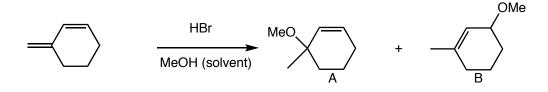
NAME

Question 5 (18 pts.) For the following reactions, provide the missing **MAJOR REACTION PRODUCT**. Indicate stereochemistry where appropriate.

- 4 -



Question 6 (25 pts.) For the following reaction, give a full curved-arrow pushing mechanism for formation of BOTH products and **indicate the Lewis acid and base at each step (LA or LB) and whether they are also Bronsted acids and bases (BA or BB).** *Include all reasonable resonance contributors for any intermediates*

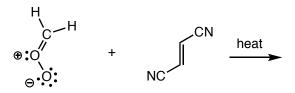


b) Indicate which is the thermodynamically controlled product and which the kinetically controlled product, and state which would be more likely to be formed at HIGHER temperature. Give a BRIEF explanation

Question 7 (25 pts.) Show how you would make the target compond on the right from the starting compound on the left. Show reagents and conditions where appropriate, and the structures of important intermediate compounds. Do not show any (arrow pushing) mechanisms.



Question 8 (12 pts) a) Give the curved arrow-pushing and the allowed product for the following cycloaddition reaction. Be sure to indicate both relative and absolute stereochemistry.



b) How many π -electrons are involved in the reaction that you drew?

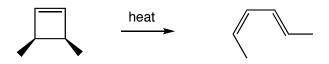
c) For the number of electrons that are involved in YOUR reaction, is a Huckel or a Mobius transition state allowed?

d) Explain WHY a Huckel/Mobius transition state (as appropriate) is allowed in YOUR reaction. Include a discussion of the energies of the electrons in the transition state in your explanation.

NAME

Question 9 (8 pts)

a) Give the curved arrow-pushing for the following electrocyclic ring opening reaction



- 6 -

b) Did **THE REACTION SHOWN** (which is not necessarily an alowed reaction) proceed via a conrotatory or a disrotatory ring closure?

c) Did **THE REACTION SHOWN** (which is not necessarily an alowed reaction) proceed via a Huckel or a Mobius transition state?

d) Is THE REACTION SHOWN allowed or forbidden?

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Question 10 (12 pts)
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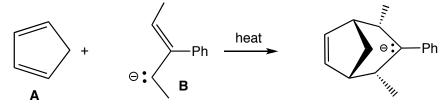
a) Give the curved arrow-pushing and the allowed product for the following electrocyclic ring closure reaction \oplus



b) ON TOP OF THE STRUCTURE, draw the HOMO of the reactant cation

c) To form YOUR product, did the reaction proceed via a conrotatory or a disrotatory ring opening?

Question 11 (12 pts.) a) Give the curved arrow-pushing for the following cycloaddition reaction



b) ON TOP OF THE STRUCTURES, draw the LUMO for A the HOMO for B

c) FOR THE PROVIDED PRODUCT (which may or not be the allowed product), was the reaction suprafacial with respect to reactant A? (Y/N) _____

d) FOR THE PROVIDED PRODUCT (which may or not be the allowed product), was the reaction suprafacial with respect to reactant B? (Y/N) _____

e) Is the REACTION SHOWN allowed or forbidden?

CHM 234, Spring 2009, Midterm #2 - 7 - NAME_____

